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CHAPTER 10 SOLUTIONS MANUAL 3. Apply How many atoms of hydrogen-1 are in a 1.007-g sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits. 1.007 g  $1 \text{ H-1 atom}$  \_\_\_ 1.672  $10^{24}$  g a. 6.023  $10^{23}$  H-1 atoms 4.

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Chapter 9 141 Section 9.1 Reactions and Equations pages 282–288  
Practice Problems pages 284–287 Write skeleton equations for the following word equations. 1. Hydrogen and bromine gases react to yield hydrogen bromide.

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